

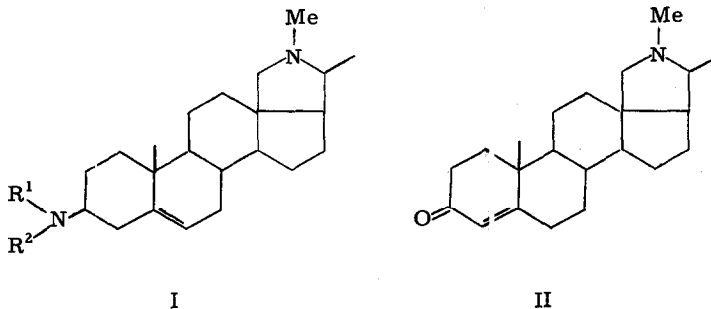
THE SYNTHESIS OF CONESSINE FROM
THE CORRESPONDING 3-KETO- Δ^4 -UNSATURATED SYSTEM

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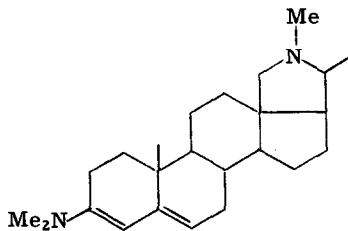
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IN connection with studies directed toward the total synthesis of conessine, I ($R^1 = R^2 = \text{Me}$), we have been in search of a facile method for its production from a 3-oxygenated derivative. The results of previous studies¹ toward this objective were not promising, e. g. reaction of dimethylamine with a 3β -p-toluenesulfonate- Δ^5 -unsaturated steroid gave, at best, very poor yields of the desired product.



¹ E. J. Corey and W. R. Hertler, J. Am. Chem. Soc. 81, 5209 (1959).



III

The selective reduction of an enamine like III to yield I ($R^1 = R^2 = \text{Me}$) was considered an attractive possibility, because the former substance promised to be readily accessible² from Δ^4 -conanene-3-one (II).³ In the present work a new direct approach to the latter ketone was developed. Conessine was converted, by the action of cyanogen bromide, into cyano-iso-conessimine, I ($R^1 = \text{Me}$, $R^2 = \text{CN}$), which, on alkaline hydrolysis, was transformed into iso-conessimine, I ($R^1 = \text{Me}$, $R^2 = \text{H}$), as previously described.⁴ This product, on treatment with N-chlorosuccinimide or t-butyl hypochlorite (to produce I, $R^1 = \text{Me}$, $R^2 = \text{Cl}$) followed by dehydrohalogenation with sodium ethoxide, then hydrolysis,⁵ afforded in 62% yield

² Cf. F. W. Heyl and M. E. Herr, J. Am. Chem. Soc. 75, 1918 (1953).

³ R. Pappo, U. S. Patent No. 2,913,455, Nov. 17, 1959; Chem. Abst. 54, 3527 (1960). In our hands the method of A. Bertho and M. Goetz, Ann. 619, 96 (1958) failed.

⁴ S. Siddiqui and R. H. Siddiqui, J. Indian Chem. Soc. 11, 787 (1934).

⁵ Cf. W. E. Bachmann, M. P. Cava and A. S. Dreiding, J. Am. Chem. Soc. 76, 5554 (1954).

Δ^4 -conanene-3-one (II), m.p. after purification 108.5-110°, $\lambda_{\text{max}}^{95\% \text{ EtOH}}$ 241 m μ (ϵ 16,000) (Found: C, 80.5%; H, 10.2%; N, 4.3%). Mixed m.p. determinations and infrared spectroscopy demonstrated the identity of this material with that of Pappo.³

Treatment of II with anhydrous dimethylamine in the presence of magnesium sulfate and a trace of *p*-toluenesulfonic acid in a sealed tube (exclusion of air) readily gave the enamine III, m.p. 151-152°, (after recrystallization from acetone) $\lambda_{\text{max}}^{\text{Et}_2\text{O}}$ 271 m μ (ϵ 18,400) (Found: C, 81.0%; H, 10.8%; N, 8.0%). Despite the known resistance of such enamine systems to reduction by lithium aluminum hydride⁶ we tried the reaction of III with sodium borohydride and, to our pleasure, found that reduction occurred smoothly and selectively to give conessine, I ($R^1 = R^2 = \text{Me}$), as the major and only isolable product.

Prior to the experiments described above we carried out a study with Δ^4 -cholestene-3-one which, under the conditions already mentioned, was converted readily into 3-dimethylamino-3,5-cholestadiene, m.p. 97-99°, $\lambda_{\text{max}}^{\text{ether}}$ 272 m μ (ϵ 19,000) (Found: C, 84.1%; H, 11.8%; N, 3.9%). Reduction with sodium borohydride afforded 3 β -dimethylamino- Δ^5 -cholestene, m.p. 151-151.5° (reported,⁷ 151°). The new synthetic

⁶ G. B. Spero, J. L. Thompson, B. J. Magerlein, A. R. Hanze, H. C. Murray, O. K. Sebek, J. A. Hogg, J. Am. Chem. Soc. **78**, 6213 (1956).

⁷ D. P. Dodgson and R. D. Haworth, J. Chem. Soc. 67 (1952).

sequence for converting a Δ^4 -3-keto-steroid into the 3β -dialkylamino- Δ^5 -unsaturated derivative thus promises to be generally applicable.

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